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## Structural and electronic properties of $\text{Ge}_n$ and $\text{EuGe}_n$ nanoclusters: A full potential DFT study

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### Abstract

In this work, the stability, structure and electronic properties of the nanoclusters of germanium ( $\text{Ge}_n$ ) and europium atom doped germanium clusters ( $\text{EuGe}_{n-1}$ ) with  $n=2$  to 12, 15 and 20 were investigated. First, the stability of nanoclusters such as  $\text{Ge}_n$  and  $\text{EuGe}_{n-1}$  was addressed using FHI-aims as a software package based on the density functional theory. Then the lowest-energy structures were selected for calculating the first vertical ionization with the symmetry adapted cluster-configuration interaction General-R (SAC-CI-General-R) method. The results of this research show that there is a good agreement between calculation and experiment ionization potential for  $\text{Ge}_n$  nanoclusters. Generally, the analyses of binding energies show that increasing the size of nanoclusters leads to more stability for nanoclusters. The most stable nanoclusters for  $\text{EuGe}_n$  can be created with exchanging the Eu atom in the most stable  $\text{Ge}_{n+1}$  nanoclusters, but there is an exception for  $n=11$  case. Here, the second difference in energy ( $\Delta_2E$ ) and gap energy are computed for the stable nanoclusters. The results of ionization energy and second difference in energy confirm that  $\text{Ge}_7$  and  $\text{Ge}_{10}$  also  $\text{EuGe}_8$  and  $\text{EuGe}_{10}$  have the most stability.

**Keywords:** germanium nanoclusters, fhi-aims, full potential, first ionization energy, second difference in energy, gap energy

For full article, refer to the Persian section